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The nuclear binding energy expressed in Joule per nucleus range from about 3.57×10^{-13} for deuterium to about 2.89×10^{-10} for uranium-238. The larger the nucleus the larger the binding energy. However, nuclear stability may be compared based on binding energy if the latter is expressed in term of Joule per nucleon.

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Mass number and atomic number are conserved in all nuclear reactions. The mass of a 25.0 g piece of ^{238}Cm (half-life: 2.4 hr) will be reduced to 14.7 g after 7.2 hr. Study Guide for Content Mastery Chemistry: Matter and Change Chapter 25

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Section 24.1 Nuclear Radiation In your textbook, read about the terms used to describe nuclear changes. Use each of the terms below just once to complete the passage. alpha particle radioactivity gamma ray radioisotope beta particles radiation X-ray radioactive decay The discovery of the

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